

Chemical Word Equations Answers

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Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead. $Zn + Pb(NO_3)_2 \rightarrow Zn(NO_3)_2 + Pb$ Single Replacement.

Word Equations Answer Key Zinc and Lead (II) nitrate react ...

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Directions: Write a balanced chemical equation for each of the word equations below. 1. aqueous sodium chloride reacts with aqueous lead (II) nitrate to yield a lead (II) chloride precipitate and aqueous sodium nitrate. sodium chloride + lead (II) nitrate \rightarrow lead (II) chloride + sodium nitrate. $NaCl + Pb(NO_3)_2 \rightarrow PbCl_2 + NaNO_3$

Chemistry: Chemical Word Equations

Title: Chemical Word Equations (Key) Author: John Bergmann & Jeff Christopherson Subject: Chemistry Created Date: 7/6/2009 7:27:55 PM

Chemical Word Equations (Key)

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WORD AND CHEMICAL EQUATIONS BALANCING WORKSHEETS WITH ANSWERS

Chemical Equation For SaltSalt is Sodium Chloride so: $NaCl$. The word equation for salt is:Sodium + Chlorine = Sodium ChlorideORThe chemical equation

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is:Na + Cl = NaCl.

Chemical equation from the word equation? - Answers

Write the word equations below as chemical equations and balance: 1) Zinc and lead (II) nitrate react to form zinc nitrate and lead. 2) Aluminum bromide and chlorine gas react to form aluminum chloride and bromine gas. 3) Sodium phosphate and calcium chloride react to form calcium phosphate and sodium chloride.

Balancing Equations Worksheet - 3-13

The chemical equation has the products on the right side, while the reactants are written on the left side. Both of them are separated by an arrow. For instance, $2H_2 + O_2 \rightarrow 2H_2O$ denotes that there are four atoms of hydrogen and 2 atoms of oxygen on both sides of the equation. The amount of reactants must be equal to the amount of products.

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A physical change in iron (nothing new gets made) B. A chemical reaction between iron and oxygen, where rust (iron oxide) is produced (made) C. A chemical reaction between iron and rust, where oxygen is produced. 3. Magnesium (a grey metal) can be burned in air to produce magnesium oxide (a white powder).which is the correct word equation for the reaction that happens?

Simple Word Equations Quiz - ProProfs Quiz

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Answer Key Writing And Balancing Chemical Equations ...

water hydrogen + oxygen (4) 7. copper(II) nitrate + water hydrogen nitrate + copper(II) hydroxide (4) 8. phosphorus + oxygen diphosphorous pentoxide (11) 9. iron + hydrochloric acid iron(II) chloride + hydrogen (2) 10. sodium hydroxide + phosphoric acid sodium phosphate + water (6)

Balancing Word Equations Chapter 9 - My Chemistry Class

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Replace immutable groups in compounds to avoid ambiguity. For example, $C_6H_5C_2H_5 + O_2 = C_6H_5OH + CO_2 + H_2O$ will not be balanced, but $XC_2H_5 + O_2 = XOH + CO_2 + H_2O$ will. Compound states [like (s) (aq) or (g)] are not required. You can use parenthesis () or brackets [].

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Word Equations You can describe a chemical reaction by writing a word equation. When silver metal is exposed to sulfur it reacts to form silver sulfide. Silver sulfide is commonly known as tarnish and turns the surface of silver objects dark and streaky black (see figure below).

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