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Geochemistry
Perspective What you
see in the rocks:

Explained by
geochemistry: By
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Tutorial 1: Calculating
a Chondrite and the
Earth

Trace Element Geochem
in GEOL209 Using
Geochemical Data
Rock and Mineral

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Identification Igneous
rock identification

Radioactive Isotopes /
Half-life Oxygen

Isotopes and the
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Earth Science

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Geological time scale
chart made easy with

tricks | memorize

geographical time

scale in 5 minutes

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~~45 Geology lecture/
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Geology Concepts

DATATION AND

GEOCHEMISTRY

Introduction to the
stable isotope Lecture

Earth and

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Major Elements

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Chapter

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Chapter 2:

Fundamental
Concepts of
Thermodynamics 23
thermodynamic
variables are derived
from them, it is worth
our while to consider

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a few of these prop-
erties. Energy is the
capacity to produce
change. It is a
fundamental property
of any system, and it
should = , : w= - w
= - M, = - = - .
2

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Thermodynamics. 20

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September 25, 2007.

Chapter 2: Energy,
Entropy and
Fundamental

Thermodynamic
Concepts. 2.1 The
Thermodynamic
Perspective. e defined
geochemistry as the
application of
chemical knowledge
and techniques to
solve geo- logical
problems.

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Page 14/41

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GEOCHEMISTRY-W.M

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Geochemistry

Chapter 11: The

Mantle and Core 487

$d(r) dr = G r^2 (r)$

$V P^2 - 4/3 V s^2$

$4 (r)r^2 dr$ 11.12

Equation 11.12

describes how density

changes in a self-

compressing, but

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otherwise uniform
sphere and is known
as the Adams-
Williamson Equation.
m) PREM

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Geochemistry
Chapter 7: Trace
Page 16/41

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Elements November
21, 2007263

Chapter 2
Solutions
typically 10^{-4} to
 10^{-12} STP cm^3/g
(10^{-1} to 10^{-9} ppm).

Their solubility in silicate melts is a strong function of pressure, as well as both atomic radius and melt composition as is illustrated in Figure 7.4.

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Chapter 5: Kinetics ©

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158 5.2.3 Reaction

Rates Consider a

reaction such as the

precipitation of

dolomite from a

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solution. Geochemistry

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Chapter 5: Kinetics C

5: K T P T

(4.5 / 5.0 – 3

customer ratings)

This book provides a

comprehensive

introduction to the

field of geochemistry.

The book first lays

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out the 'geochemical toolbox': the basic principles and techniques of modern geochemistry, beginning with a review of thermodynamics and kinetics as they apply to the Earth and its environs.

William M. White

Page 20/41

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Geochemistry –

World of Digitals

Chapter 2
William White

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teaches geochemistry

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University of

California, Berkeley

and a PhD in...

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Chapter 8: Radiogenic
Isotope Geochemistry

320 January 10,

2001 also binds

quarks together to

form hadrons, a class

of particles that in-

cludes neutrons and

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protons. The intensity of the strong force decreases rapidly with distance, so that at distances more than about 10-14 m it is weaker than the electromagnetic force.

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Chapter 8: Radiogenic
Isotope ...

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In this chapter we will consider the behavior of trace elements, particularly in magmas, and introduce methods to model this behavior. Though trace elements, by definition, constitute only a small fraction of a system of interest, they provide geochemical and

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geological
information out of
proportion to their
abundance.

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Chapter 7: Trace
Elements Chapter ...

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Chapter 2:
Fundamental

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Concepts of Chemistry

Thermodynamics 24

Chapter 2
September 26, 2001

Solutions
As all other

thermodynamic variables are derived from them, it is worth our while to consider a few of these properties. Energy is the capacity to produce change. It is a fundamental property of any

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system, and it should
be familiar from
physics.

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Chapter 10:

Cosmochemistry

Page 27/41

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418July31,206 we learn about the evolution of the Earth by examining old rocks, we can learn about the evolution of the cosmos by looking at old stars. The old stars of Population II are considerably poorer in heavy elements than are young stars.

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Cosmochemistry ...

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Chapter 4:

Applications of

Thermodynamics 120

October 17, 2001

$m = m + + 22 \ 2 \ o \ RT \ X$

W In 4.15 G Equation

4.14 is Raoult ' s

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Law, letting: $\mu^* = \mu^\circ + W G$ or $W G = RT \ln h$ then 4.15 is Henry ' s Law. Thus the interaction parameter can be related to the parameters of Henry ' s Law, and activity coefficient. In the Mar-

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Applications of ...

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Chapter 3: Solutions

William White

teaches geochemistry
as a Professor of
earth and

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He received a B.A. in geology from the University of California, Berkeley and a PhD in oceanography from the University of Rhode Island. William M. White

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Page 32/41

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Chapter 7: Trace

Elements 261 HCO₃⁻

, Mg²⁺, Ca²⁺, K⁺
and Na⁺ (and H₂O, of
course) can be

considered a trace
constituent, though

Sr²⁺, HBO₃⁻, and

Br⁻ are sometimes

considered major

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constituents also
(constituents or
species is a better
term here than
elements).

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Chapter 8: Radiogenic Isotope Geochemistry

320 January 10,

2001 also binds

quarks together to

form hadrons, a class

of particles that in-

cludes neutrons and

protons.

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Page 35/41

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Chapter 7: Trace

Elements. W. M.

White Geochemistry.

Chapter 7: Trace

Elements. November

3, 2009260 HCO. 3

- , Mg²⁺, Ca²⁺,

K⁺and Na⁺(and H₂O,

of course) can be

considered a trace

constituent, though

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Sr^{2+} , HCO_3^- , and Br^- are sometimes considered major constituents also (constituents or species is a better term here than elements).

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Chapter 7: Trace
Elements

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W. M. White Chapter
9: Stable Isotopes.

Geochemistry 9.2.1.1
The Quantum

Mechanical Origin of
Isotopic

Fractionations. It is
fairly easy to
understand, at a
qualitative level at
least, how some
isotope fractionations
can arise from
vibrational motion.

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Chapter 9: Stable

Isotopes ...

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Elements. 259.

Chapter 7: Trace

Elements in Igneous

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Processes. 7.1

INTRODUCTION. In this chapter we will consider the behavior of trace elements, particularly in magmas, and introduce methods to model this behavior. Though trace elements, by definition, constitute only a ...

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